Bioactive compounds and volatile compounds of Thai bael fruit (Aegle marmelos (L.) Correa) as a valuable source for functional food ingredients

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Abstract: Bael fruit (*Aegle marmelos* (L.) Correa) is an attractive and characteristically sweet aroma fruit, native to Southeast Asia, known to be a good source of natural antioxidants and bioactive compounds. In this study, the results showed that fully ripe Thai bael fruit pulps had total, soluble, and insoluble dietary fiber contents of 19.84, 11.22, and 8.62 g/ 100 g dry weight (dw), respectively. Determination of antioxidant activities by 2-diphenyl-1-picryhydrazyl (DPPH) free radical scavenging and ferric reducing antioxidant power (FRAP) assays resulted in 6.21 µg dw/ µg DPPH and 102.74 µM trolox equivalent (TE)/ g dw, respectively. It was also found to have total phenolic, total flavonoid, total carotenoid, and ascorbic acid contents of 87.34 mg gallic acid equivalent (GAE)/ g dw, 15.20 mg catechin equivalent (CE) / g dw, 32.98 µg / g dw, and 26.17 mg / 100 g dw, respectively. Volatile compounds in bael fruit pulp were analyzed using the solid-phase microextraction (SPME) / gas chromatography (GC) / mass spectrometry (MS) method. A total of 28 volatile compounds were identified, and the dominant components were monoterpenes and sesquiterpenes. Among these components, limonene was the major constituent producing the characteristic bael fruit flavor.

Keywords: Bael fruit, bioactive compounds, antioxidant activities, volatile compounds, functional food

INTRODUCTION

Bael fruit (Aegle marmelos (L.) Correa) is a tropical fruit native to Southeast Asia and belongs to the Rutaceae family. It is grown throughout India as well as in Sri Lanka, Pakistan, Bangladesh, Burma, Thailand, and most of the Southeast Asian countries (Singh and Roy, 1984). In Thailand, it is commonly found growing in many regions, especially the lower north and central part consisting of Phichit, Prachin Buri, and Phitsanulok provinces. There are no standard names for bael fruit cultivars. The general cultivars known in Thailand are locally "Matoom Kai" (in Thai language "Matoom" means bael fruit and "Kai" refers to cultivar). It is usually used for household consumption and traditional medicine, but some are planted for trade in particular regions (Subhadrabandhu, 2001).

The peel of the fruit which is a very hard shell and green to brown in color depends on ripening stage. The appearance of yellow or orange edible pulp is like a boiled pumpkin, possesses a slightly sweet taste and a characteristic floral, terpene-like aroma, very fragrant and pleasantly flavored. Seeds are surrounded by slimy transparent mucilage. The bael fruit pulp contains many functional and bioactive compounds such as carotenoids, phenolics, alkaloids, coumarins, flavonoids, terpenoids, and other antioxidants which may protect us against chronic diseases. Total dietary fiber found in this fruit can be divided into insoluble dietary and soluble dietary fiber (mucilage and pectin). In addition, it also contains many vitamins and minerals including vitamin C, vitamin A, thiamine, riboflavin, niacin, calcium, and phosphorus (Dikshit and Dutt, 1930; Parmar and Kaushal, 1982; Roy and Khurdiya, 1995). Therefore, bael fruit may indicate that it is one of the important plants used for indigenous traditional medicine. There are innumerable references of its uses in traditional medicine (Arseculeratne et al., 1981; Karunanayake et al., 1984; Singh, 1986; Nagaraju and Rao, 1990).

The uses of bael fruit in aspects of food have many forms in each country. For example, the ripe fruit is consumed fresh and also prepared as nectar, squash, sherbet, jam, marmalade, and cream in India (Morton, 1987). However, in Thailand these fruits are usually cut into pieces and dried, packed in bags or pulverized and packed as tea bags, and preserved in syrup as the bael fruit glacé which is normally used as dessert or an ingredient for cakes.

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Nowadays, the world market for functional foods and nutraceuticals is large and growing. According to the literature, the characteristic of Thai bael fruit in terms of bioactive compounds and characteristic flavor was considered to have a potential for use as functional food and value added processed products. There is considerable amount of literature that examines the bioactive compounds and volatile compounds in fruits such as guava, mango, and berries. However, there have been no data on bioactive and volatile compounds of Thai bael fruit. Therefore, the purpose of this study was to determine the bioactive compounds and volatile compounds in Thai bael fruit (Matoom Kai), a cultivar usually found and consumed in Thailand, and also to investigate the possibilities of utilizing Thai bael fruit as a valuable functional food ingredient or a resource for nutraceutical products in the future. Bael fruit is from a perennial plant which is present in Thai way of life from ancient times, but today this plant has some limits in its use and the risk of extinction. The results of this study can add to the scientific literature and useful application from bael fruit for producers and consumers to be aware of the importance and utilization of this useful resource.

MATERIALS AND METHODS

Plant materials

Fresh Thai bael fruits *Aegle marmelos* (L.) Correa (Matoom Kai) were obtained from a bael fruit garden in Phichit, Thailand. Fully ripe bael fruits (Figure 1) were chosen for this study because this is what usually consumed fresh and used as a main ingredient in many food products. The pulps were removed from the fruits, and were analyzed for physicochemical properties, bioactive compounds, and volatile compounds. All chemicals and solvents used in this experiment were of analytical grade,

purchased from Sigma Chemical Co., Ltd (St. Louis, MO, USA) and Sigma Aldrich Co., Ltd (Steinheim, Germany). All analyses was performed in triplicate.

Determination of physicochemical properties

The pulps of bael fruit were analyzed for total soluble solids by using a hand-held refractometer (Atago 32-62°Brix, Japan), pH was measured using a digital pH meter (Eutech, Cyber Scan pH 1000 Bench, Singapore), reducing sugars were determined as glucose by using Nelson-Somogyi method (Nelson, 1944), total acidity and moisture content were analyzed according to standard AOAC methods (AOAC, 1995). The color of fresh bael fruit pulps was measured using a chromameter (Minolta, CR400) with reference to illuminant D_{65} , expressed with the value of Hunter. The L* value is a measurement of lightness and varies from 0 (black) to 100 (white); the a* value varies from -a* (green) to +a* (red); the b* value varies from -b* (blue) to +b* (yellow); C* (chroma or color saturation); and °h (hue angle).

Determination of dietary fiber

Total, soluble, and insoluble dietary fiber (TDF, SDF, and IDF) were analyzed according to standard AOAC methods (AOAC, 1995).

Antioxidant activity determination

Samples were prepared by the method modified from Velioglu *et al.* (1998). Fresh bael fruit pulps were ground and extracted with 95% ethanol in the dark at 25°C for 4.5 h. The free radical-scavenging activity of sample extracts was evaluated using the DPPH assay, according to the methods of Masuda *et al.* (1999) and Maisuthisakul *et al.* (2007). The percentage of DPPH radical-scavenging activity was calculated and plotted against the sample extract concentration (μ g/ ml) to determine the amount of extract necessary to decrease DPPH radical



Figure 1: Physical characteristics of "Matoom Kai": (a) outer hard shell skin and (b) internal pulp

concentration by 50% (called EC_{50}). The unit of EC_{50} was later converted to µg dry weight (dw) / µg DPPH. FRAP assay was done according to Benzie and Strain (1996) with some modifications. Briefly, the fruit extracts (50 µl) were allowed to react with 950 µl of the FRAP solution for 4 min in dark conditions. The standard curve was linear between 82 and 625 µM trolox. Results were expressed in µM trolox equivalent (TE) / g dw.

Determination of total phenolic and total flavonoid content

Samples were prepared in the same way as for the antioxidant activity determination. Total phenolic content was determined by the Folin-Ciocalteu assay (Waterhouse, 2005). Gallic acid (50-500 mg/l) was used for calibration of the standard curve. The results were expressed as mg gallic acid equivalent (GAE) / g dw.

Total flavonoid content was measured by the aluminum chloride colorimetric assay (Zhishen *et al.*, 1999). Measurements were calibrated to a standard curve of prepared catechin solution (20-100 mg/l). The results were expressed as mg catechin equivalent (CE)/g dw.

Analysis of total carotenoids and ascorbic acid

Total carotenoids were determined according to Gross (1991); Talcott and Howard (1999). The absorbance at 470 nm of extracts in acetone/ ethanol (1:1) with 200 mg/L butylated hydroxytoluene (BHT) was measured. The results were expressed in µg carotenoid/ g dw.

Ascorbic acid was determined based on the quantitative discoloration of 2,6dichlorophenolindophenol (DCPIP), according to Pearson's (1976) method. The samples were extracted in 0.4% oxalic acid solution, and compared with standard ascorbic acid.

Volatile compounds analysis (modified from Chen et al., 2006)

Volatile compounds were isolated by using solidphase microextraction (SPME) method. The volatiles were sampled by manual headspace solid phase microextraction at 60°C. The fibre (100 µm PDMS, Supelco) was pierced into the injection port of the GC/MS after 20 min of sampling, and then desorbed at 200°C for 5 min. Gas chromatography/ mass spectrometry condition: An Agilent 6890 GC equipped with an Agilent 5973 mass-selective detector (Agilent Technologies) was used, with the injector and detector maintained at 200 and 260°C, respectively. The column (HP-Innowax) dimensions were 0.25 mm i.d. × 30 m × 0.25 µm film thickness. The carrier gas (He) had a flow rate of 5.7 ml/ min. The temperature program was isothermal at 50°C for 10 min, increase to 240°C at 15°C/ min, then held for 10 min. The mass spectrometer was operated in the electron impact (EI) mode with an electron energy of 70 eV; ion source and quadrupole maintained at 230 and 150°C, respectively; mass range m/z 10-350. Compounds were identified by matching mass spectra (quality match >80%) and retention indices with the Chemstation Wiley Spectral Library of standard compounds, as well as from NIST, Mass Finder, and the literature (Tressl *et al.*, 1982; Davies, 1990; Chung *et al.*, 1993; Píry *et al.*, 1995; Kocsis *et al.*, 2002; Choi, 2003; Högnadóttir and Russell, 2003; Mau *et al.*, 2003; Culleré *et al.*, 2004; Szafranek *et al.*, 2005).

RESULTS AND DISCUSSION

Physicochemical properties

Bael fruit pulps were analyzed for total soluble solids, pH, reducing sugar, total acidity, moisture content, and internal color. The results are presented in Table 1. Total soluble solids of the pulps were 39.50°Brix, about twice as high as in most other fruits. The bael fruit pulp had pH, reducing sugar, total acidity, and moisture content of 5.37, 39.60 mg glucose/g fresh weight (fw), 0.94%, and 67.74%, respectively. For the physicochemical properties between Thai and Indian bael fruits for different cultivars, the results showed that pH, reducing sugar, and moisture content of Thai bael fruits were similar to Indian bael fruit. However, total soluble solids and total acidity of Thai bael fruit were higher than Indian bael fruit. The color of fruit in this study was orange and yellow corresponding to L*a*b* measurement, and similar to the internal color of Indian bael fruit pulp (various shades of yellow to orange and yellow) reported by Roy and Khurdiya (1995).

Bioactive compounds

Bioactive compounds found in bael fruit pulp are presented in Table 2. The results showed that bael fruit pulps had TDF, SDF, and IDF contents of 19.84, 11.22, and 8.62 g/ 100 g dw, respectively. The results indicate that the fruits are relatively rich in dietary fiber because they were in range of fruits which are defined as high dietary fiber fruits such as watermelon, orange, banana, mango, papaya, pineapple, and apple (TDF 9-24, SDF 4-8, and IDF 6-17 g/ 100 g dw) reported by Ramulu and Rao (2003).

Antioxidant activities were measured in ethanol extract obtained using DPPH and FRAP assays. The DPPH assay was selected to evaluate the

Physicochemical characteristics	Mean±SD
Total soluble solids (°Brix)	39.50 ± 0.87
pH	5.37 ± 0.03
Reducing sugar (mg glucose/ g fw ^a)	39.60 ± 0.89
Total acidity (%)	0.94 ± 0.04
Moisture (%)	67.74 ± 0.25
Internal color	
L*	40.11 ± 0.27
a*	7.82 ± 0.20
b*	35.84 ± 0.24
C*	36.68 ± 0.23
°h	77.70 ± 0.33

Table 1: Physicochemical properties of "Matoom Kai"

^a fw = fresh weight basis, All values were performed in triplicate.

Bioactive compounds	Mean±SD	
Total dietary fiber (TDF) (g/ 100 g dw ^a)	19.84 ± 0.01	
Soluble dietary fiber (SDF)	11.22 ± 0.06	
Insoluble dietary fiber (IDF)	8.62 ± 0.04	
Antioxidant activities		
DPPH assay (EC ₅₀ , µg dw/ µg DPPH)	6.21 ± 0.34	
FRAP assay ($\mu M TE^{b}/g dw$)	102.74 ± 3.01	
Total phenolics (mg GAE ^{c} / g dw)	87.34 ± 4.44	
Total flavonoids (mg CE ^d / g dw)	15.20 ± 0.51	
Total carotenoids (µg/ g dw)	32.98 ± 0.51	
Ascorbic acid (mg/ 100 g dw)	26.17 ± 0.85	

Table 2: Bioactive compounds of "Matoom Kai"

^a dw = dry weight basis, ^b TE = trolox equivalent

^c GAE = gallic acid equivalent, ^d CE = catechin equivalent

All values were performed in triplicate.

antioxidant activities of bael fruit extracts in terms of EC_{50} , the antioxidant activities obtained from this assay (6.21 μ g dw/ μ g DPPH) were comparable to that of fruits, vegetables, herbs, and chewing plants such as Diospyros kaki L., Garcinia mangostana L., Spondias pinnata Kurz, Leucaena glauca Benth, and Piper betel Linn. (0.3-7 µg dw/ µg DPPH) which contain high antioxidant activity reported by Maisuthisakul et al. (2007). The antioxidant activities as determined by FRAP assay of bael fruit $(102.74 \ \mu M \ TE/g \ dw)$ were comparable to herbs and vegetables such as roselle, Asian pennywort, Thai basil, beet, and sweet potato (86-150 μ M TE/g dw) reported by Ou et al. (2002) and Wong et al. (2006). Therefore, bael fruit is another fruit that has high antioxidant activities.

It can be indicated that the major antioxidants in bael fruit are phenolics, flavonoids, carotenoids, and vitamin C (Morton, 1987; Roy and Khurdiya, 1995). The results showed that bael fruit pulps had total phenolic content of 87.34 mg GAE/gdw. Total phenolics in bael fruits were in the range of traditional Chinese medicinal plants associated with anticancer (2.2-503 mg GAE/g dw), as well as higher than for common fruits and vegetables including kiwifruit, orange, pear, garlic, carrot, and spinach (1.2-10.8 mg GAE/ g dw) reported by Cai et al. (2004). The results for total flavonoid content showed that bael fruit pulps had relatively high total flavonoids (15.20 mg CE/ g dw) when compared with medicinal plants such as Alcea kurdica, Stachys lavandulifolium, Valeriana officinalis, Lavandula

officinalis, and Melissa officinalis (0.22-10 mg CE/g dw) reported by Bouayed et al. (2007). In addition, total flavonoids in bael fruit were also higher than for fruits and vegetables including apple, raspberry, peach, broccoli, celery, and tomato (2.5-190.3 mg CE/100 g fw) reported by Marinova et al. (2005). In the case of total carotenoids, bael fruit pulps had total carotenoid content of $32.98 \,\mu\text{g}/\text{g}$ dw which is in the range for fruits and vegetables such as papaya, mango, pumpkin, ginger, and cabbage (12-70 µg carotenoid/g dw) reported by Inocent et al. (2007) and Kandlakunta et al. (2008). The ascorbic acid content of bael fruit pulps (26.17 mg/ 100 g dw) were also in the range of some varieties of herb and vegetable leaves (Leucaena glauca Benth and Ocimum basilicum Linn.) as well as berry and fruit seeds (Nephelium lappaceum Linn., Parkia speciosa Hassk., and Tamarindus indica Linn) (19.3-39.8 mg/ 100 g dw) reported by Maisuthisakul et al. (2008).

Volatile compounds

Twenty-eight compounds were characterized by SPME/GC/MS (Figure 2 and Table 3). Among these components, monoterpenes and sesquiterpenes such as Limonene, p-Cymene, β -Phellandrene, and Dihydro- β -Ionone were recognized as the main components. These compounds are important contributors to fruit aroma showed in Table 4. In this study, the most important volatile compound in bael fruit pulp seems to be limonene because this

compound had the highest percentage of peak area in chromatogram.

When the volatile compounds of Thai bael fruit and Sri Lankan bael fruit reported by MacLeod and Pieris (1981) are compared, the results showed that monoterpenes and sesquiterpenes seem to be the main volatile constituents of both Thai and Sri Lankan bael fruit. Among these components, limonene was one of the major constituents that produces the characteristic bael fruit flavor. Tokitomo et al. (1982) reported that Limonene, Δ^3 -Carene, p-Cymene, Linalool oxide, Linalool, Benzyl acetate, Benzyl alcohol, 2-Phenyleyhanol, and β -Ionone are important constituents to produce the basic or fundamental flavor quality of bael fruit from Sri Lanka, and 3,7-dimethyl-1,5,7-octatrien-3-ol and its isomer are responsible for making the aroma of this fruit more attractive. It is interesting that the major components of bael fruit flavor were monoterpenes and sesquiterpenes, which is similar to citrus fruits of the Rutaceae family reported by Sawamura et al. (1990), Sawamura et al. (1991) and Choi (2003), especially limonene. Limonene is one of the bioactive compounds that provide health benefit in terms of anticancer (Bidlack, 2000; Shi et al., 2002). Although the appearance of the fruit is like wood apple or quince, the aroma components are quite different from those of wood apple and quince whose aroma consists mainly of various esters (Schreyen et al., 1979; MacLeod and Pieris, 1981).



Figure 2: Chromatogram of volatile compounds identified from SPME/GC/MS

Peak No.	RI^{a}	Compound	% Area
1	1093	Hexanal	0.93
2	1147	Isoamyl acetate	2.13
3	1202	Limonene	32.48
4	1217	β-Phellandrene	3.13
5	1279	p-Cymene	27.19
6	1295	Acetoin	0.70
7	1379	(E)-2-Octenal	0.60
8	1401	(E,E)-2,4-Heptadienal	1.08
9	1414	Dehydro-p-cymene	1.47
10	1425	Linalool oxide	0.88
11	1432	3,5-Octadiene-2-one	1.44
12	1463	α-Cubebene	1.75
13	1470	trans-p-Mentha-2,8-dienol	0.71
14	1488	Citronellal	1.42
15	1558	β-Cubebene	1.18
16	1594	β-Caryophyllene	1.88
17	1600	Hexadecane	1.71
18	1665	Pulegone	1.71
19	1680	α-Humulene	1.08
20	1729	Verbenone	1.25
21	1751	Carvone	0.98
22	1759	Carvyl acetate	0.82
23	1825	Dihydro-β-Ionone	3.95
24	1840	(E)-6,10-dimethyl-5,9-Undecadien-2-one	2.16
25	1947	β-Ionone	2.95
26	1999	Caryophyllene oxide	2.57
27	2015	Humulene oxide	0.78
28	2860	Hexadecanoic acid	0.92

Table 3: Volatile compounds identified from SPME/GC/MS

^a RI = Kovat's retention indices

Table 4: Examples of dominant volatile compounds of "Matoom Kai", their chemical groups and other sources

Chemical groups	Volatile compounds	Sources	
monoterpene	Limonene	orange, mango, grape, lemon,	
	β-Phellandrene	blackcurrant	
	p-Cymene	mango, citrus, lime oil	
	Linalool oxide	orange, berry	
	Pulegone	blackberry	
	β-Ionone	raspberry, mango	
	Dihydro-β-Ionone	raspberry	
sesquiterpene	β-Cubebene	citrus	
	β-Caryophyllene	lemon, guava, mango, grape, blackcurrant	

References: MacLeod and Snyder (1985), Maarse (1991), Sawamura *et al.* (1991), Píry *et al.* (1995), Pino *et al.* (2005), Wenguang *et al.* (2007), Harb *et al.* (2008)

CONCLUSION

The physicochemical properties of Thai bael fruit are similar to Indian bael fruit. Bioactive compounds of bael fruit in this study contain relatively high content of dietary fiber, ascorbic acid, total phenolics, total flavonoids, total carotenoids, and also strong antioxidant activities. The volatile compounds of Thai bael fruit were comparable to bael fruit from Sri Lanka. The main components were monoterpenes and sesquiterpenes. Among these components, limonene was the major constituent producing the characteristic bael fruit flavor. This compound was also found in citrus fruits. This study indicates that Thai bael fruit may impart health benefits when it is used in functional food products and should also be regarded as a potential nutraceutical resource in the future. In addition, it can be used as food additive because of its typical color, flavor, and texture. These results are useful for developing and improving the quality of bael fruit cultivar in order to provide more value addition and usefulness from bael fruit.

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REFERENCES

- AOAC. 1995. Official Methods of Analysis of the AOAC International. Association of Official Analytical Chemists. Washington, D. C.
- Arseculeratne, S. N., Gunatilaka, A. A. L. and Panabokke, R. G. 1981. Studies on medicinal plants of Srilanka: occurrence of pyrolizidine alkaloids and hepatotoxic properties in some traditional medicinal herbs. Journal of Ethnopharmacology 4: 159-177.
- Benzie, I. F. F. and Strain, J. J. 1996. The ferric reducing ability of plasma (FRAP) as a measure of "Antioxidant Power" the FRAP assay. Analytical Biochemistry 239: 70-76.
- Bidlack, W. R. 2000. Phytochemicals as Bioactive Agents. Lancaster: Technomic.
- Bouayed, J., Piri, K., Rammal, H., Dicko, A., Desor, F., Younos, C. and Soulimani, R. 2007. Comparative

evaluation of the antioxidant potential of some Iranian medicinal plants. Food Chemistry 104: 364–368.

- Cai, Y., Luo, Q., Sun, M. and Corke, H. 2004. Antioxidant activity and phenolic compounds of 112 traditional Chinese medicinal plants associated with anticancer. Life Sciences 74: 2157-2184.
- Chen, J. L., Yan, S., Feng, Z., Xiao, L. and Hu, X. S. 2006. Changes in the volatile compounds and chemical and physical properties of Yali pear (*Pyrus bertschneideri* Reld) during storage. Food Chemistry 97: 248-255.
- Choi, H. S. 2003. Character impact odorants of *Citrus* Hallabong [(*C. unshiu* Marcov x *C. sinensis* Osbeck) x *C. reticulata* Blanco] cold-pressed peel oil. Journal of Agricultural and Food Chemistry 51: 2687-2692.
- Chung, T.Y., Eiserich, J. P. and Shibamoto, T. 1993. Volatile compounds isolated from edible Korean chamchwi (Aster scaber Thunb). Journal of Agricultural and Food Chemistry 41: 1693-1697.
- Culleré, L., Escudero, A., Cacho, J. and Ferreira, V. 2004. Gas chromatography-olfactometry and chemical quantitative study of the aroma of six premium quality Spanish aged red wines. Journal of Agricultural and Food Chemistry 52: 1653-1660.
- Davies, N. W. 1990. Gas chromatographic retention indices of monoterpenes and sesquiterpenes on methyl silicone and carbowax 20M phases. Journal of Chromatography A 503:1-24.
- Dikshit, B. B. L. and Dutt S. 1930. Preliminary chemical examination of *Aegle marmelos* or the Indian Bel. Journal of the Indian Chemical Society 7: 759-764.
- Gross, J. 1991. Pigments in Vegetables: Chlorophylls and Carotenoids. New York: Van Nostrand Reinhold.
- Harb, J., Bisharat, R. and Streif, J. 2008. Changes in volatile constituents of blackcurrants (*Ribes nigrum* L. cv. 'Titania') following controlled atmosphere storage. Postharvest Biology and Technology 47: 271–279.
- Högnadóttir, A. and Russell, R.L. 2003. Identification of aroma active compounds in orange essence oil using gas chromatography - olfactometry and gas chromatography - mass spectrometry. Journal of Chromatography A 998: 201-211.
- Inocent, G., Ejoh, R. A., Issa, T. S., Schweigert, F. J. and Tchouanguep, M. F. 2007. Carotenoids content of

some locally consumed fruits and yams in Cameroon. Pakistan Journal of Nutrition 6(5): 497-501.

- Kandlakunta, B., Rajendran, A. and Thingnganing, L. 2008. Carotene content of some common (cereals, pulses, vegetables, spices and condiments) and unconventional sources of plant origin. Food Chemistry 106: 85-89.
- Karunanayake, E. H., Welihinda, J., Sirimanne, S. R. and Sinnadorai, G. 1984. Oral hypoglycaemic activity of some medicinal plants of Srilanka. Journal of Ethnopharmacology 11: 223-231.
- Kocsis, N., Amtmann, M., Mednyánszky, Z. and Korány, K. 2002. GC-MS investigation of the aroma compounds of Hungarian Red Paprika (*Capsicum annuum*) cultivars. Journal of Food Composition and Analysis 15:195-203.
- Maarse, H. 1991. Volatile Compounds in Foods and Beverages. New York: Marcel Dekker.
- MacLeod, A. J. and Pieris, N. M. 1981. Volatile flavor components of beli fruit (*Aegle marmelos*) and a processed product. Journal of Agricultural and Food Chemistry 29: 1262-1264.
- MacLeod, A. J. and Pieris, N. M. 1981. Volatile flavor components of wood apple (*Feronia limonia*) and a processed product. Journal of Agricultural and Food Chemistry 29: 49-53.
- MacLeod, A. and Snyder, C. H. 1985. Volatile components of two cultivars of mango from Florida. Journal of Agricultural and Food Chemistry 33: 380.
- Maisuthisakul, P., Suttajit, M. and Pongsawatmanit, R. 2007. Assessment of phenolic content and free radical-scavenging capacity of some Thai indigenous plants. Food Chemistry 100: 1409–1418.
- Maisuthisakul, P., Pasuk, S. and Ritthiruangdej, P. 2008. Relationship between antioxidant properties and chemical composition of some Thai plants. Journal of Food Composition and Analysis 21: 229-240.
- Marinova, D., Ribarova, F. and Atanassova, M. 2005. Total phenolics and total flavonoids in Bulgarian fruits and vegetables. Journal of the University of Chemical Technology and Metallurgy 40 (3): 255-260.
- Masuda, T., Yonemori, S., Oyama, Y., Takeda, Y., Tanaka, T., Andoh, T., Shinobara, A. and Nakata, M. 1999. Evaluation of the antioxidant activity of

environmental plants: activity of the leaf extracts from seashore plants. Journal of Agricultural and Food Chemistry 47: 1749-1754.

- Mau, J. L., Ko, P. T. and Chyau, C. C. 2003. Aroma characterization and antioxidant activity of supercritical carbon dioxide extracts from *Terminalia catappa* leaves. Food Research International 36: 97-104.
- Morton. J. F. 1987. Fruits of Warm Climates. Winterville, N.C.: Creative Resource Systems.
- Nagaraju, N. and Rao, K. N. 1990. A survey of plant crude drugs of Rayalaseema, Andhra Pradesh Indian. Journal of Ethnopharmacology 29: 137-158.
- Nelson, N. 1944. A photometric adaptation of the Somogyi method for determination of glucose. Journal of Biological Chemistry 153: 375-380.
- Ou, B., Huang, D., Hampsch-Woodill, M., Flanagan, J. A. and Deemer, E. K. 2002. Analysis of antioxidant activities of common vegetables employing oxygen radical absorbance capacity (ORAC) and ferric reducing antioxidant power (FRAP) assays: a comparative study. Journal of Agricultural and Food Chemistry 50: 3122-3128.
- Parmar, C. and Kaushal, M. K. 1982. Wild Fruits of the Sub-Himalayan Region. New Delhi: Kalyani Publishers.
- Pearson, D. 1976. The chemical Analysis of Foods. Edinburgh: Churchill Livingstone.
- Pino, J. A., Mesa, J., Mu oz, Y., Martí, M. P. and Marbot, R. 2005. Volatile components from mango (*Mangifera indica* L.) cultivars. Journal of Agricultural and Food Chemistry 53: 2213-2223.
- Píry, J., Príbela, A., Ďurčanská, J. and Farkaš, P. 1995. Fractionation of volatiles from blackcurrant (*Ribes nigrum* L.) by different extractive methods. Food Chemistry 54: 73-77.
- Ramulu, P. and Rao, P. U. 2003. Total, insoluble, and soluble dietary fiber contents of Indian fruits. Journal of Food Composition and Analysis 16: 677-685.
- Roy, S. K. and Khurdiya, D. S. 1995. Other subtropical fruit. In Salunkhe, D. K. and Kadam, S. S. (eds.), Handbook of Fruit Science and Technology: Production, Composition, Storage and Processing. New York: Marcel Dekker.

- Sawamura, M., Kuwahara, S., Shichiri, K. and Aoki, T. 1990. Volatile constituents of several varieties of pummelos and a comparison of the nootkatone levels in pummelos and other citrus fruits. Journal of Agricultural and Biological Chemistry 54 (3): 803-805.
- Sawamura, M., Shichiri, K., Ootani, Y. and Zheng, X. H. 1991. Volatile constituents of several varieties of pummelos and characteristics among citrus species. Journal of Agricultural and Biological Chemistry 55 (10): 2571-2578.
- Schreyen, L., Dirinch, P., Sandra, P. and Schamp, N. 1979. Flavor analysis of quince. Journal of Agricultural and Food Chemistry 27: 872-876.
- Shi, J., Mazza, G. and Maguer, M. L. 2002. Functional Foods: Biochemical and Processing Aspects. Florida: CRC Press.
- Singh, R. N. and Roy, S. K. 1984. The Bael: Cultivation and Processing. New Delhi: ICAR.
- Singh, Y. N. 1986. Traditional medicine in Fiji: Some herbal folk cures used by Fiji Indians. Journal of Ethnopharmacology 15: 57-88.
- Subhadrabandhu, S. 2001. Under-Utilized Tropical Fruits of Thailand. Food and Agriculture Organization of the United Nations Regional Office for Asia and the Pacific, Bangkok, Thailand.
- Szafranek, B., Chrapkowska, K., Pawi ska, M. and Szafranek, J. 2005. Analysis of leaf surface sesquiterpenes in potato varieties. Journal of Agricultural and Food Chemistry 53: 2817-2822.
- Talcott, S. T. and Howard, L. R. 1999. Phenolic autooxidation is responsible for color degradation in processed carrot puree. Journal of Agricultural and Food Chemistry 47: 2109-2115.

- Tokitomo, Y., Shimono, Y., Kobayashi, A. and Yamanishi, T. 1982. Aroma components of bael fruit (*Aegle marmelos* CORREA). Journal of Agricultural and Biological Chemistry 46 (7): 1873-1877.
- Tressl, R., Bahri, D. and Engel, K. H. 1982. Formation of eight-carbon and ten-carbon components in mushrooms (*Agaricus campestris*). Journal of Agricultural and Food Chemistry 30:89-93.
- Velioglu, Y. S., Massa, G., Gao, L. and Oomah, B. D. 1998. Antioxidant activity and total phenolics in selected fruits, vegetables, and grain products. Journal of Agricultural and Food Chemistry 46: 4113-4117.
- Waterhouse, A. L. 2005. Determination of total phenolics. In Wrolstad R. E., Acree, T. E., Decker, E. A., Penner, M. H., Reid, D. S., Schwartz, S. J., Shoemaker, C. F., Smith, D. and Sporns, P. (eds.), Handbook of Food Analytical Chemistry: Pigments, Colorants, Flavors, Texture, and Bioactive Food Components. New Jersey: John Wiley and Sons.
- Wenguang, J., Wenlai, F., Yan, X., Guang'ao, Z., Jiming, L. and Ying, Y. 2007. Analysis of free terpenoids in *Vitis vinifera* using solvent assisted flavour evaporation and gas chromatography-tandem mass spectrometry. Chinese Journal of Chromatography 25 (6): 881–886.
- Wong, S. P., Leong, L. P. and Koh, J. H. W. 2006. Antioxidant activities of aqueous extracts of selected plants. Food Chemistry 99: 775-783.
- Zhishen, J., Mengcheng, T. and Jianming, W. 1999. The determination of flavonoid contents in mulberry and their scavenging effects on superoxide radicals. Food Chemistry 64: 555-559.